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## Usefulness of Reference Materials in Calibration of Enzyme Activities<sup>1)</sup>

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**Summary:**  $\alpha$ -Amylase, alkaline phosphatase and  $\gamma$ -glutamyltransferase were studied in a multicentre evaluation. Analyses were performed on different patient samples. Each enzyme was assayed in two different laboratories at both 30 and 37 °C, with widely used reagent kits and with the IFCC reference method (if in existence). Results differed considerably according to the measurement procedure. Data also showed that it was not possible to employ a constant conversion factor for one enzyme and different techniques between 30 and 37 °C.

Calibration with three reference materials extensively improved the intermethod consistency for most of the tested measurement procedures. It was possible to transfer accuracy from the method used for the certification of the reference material to routine procedures, by using the reference material as calibrator. Temperature did not seem to be a crucial variable for the implementation of the enzyme calibrator approach.

### Introduction

The method-dependency of the results of enzymic activity determinations presents problems of interpretation for clinicians. The aims of the present study were to evaluate:

- the degree of method-dependency of results obtained using routine conditions
- the effect of the use of appropriate enzyme calibrators on the improvement of between-method coherency.

This preliminary study was limited to  $\alpha$ -amylase (EC 3.2.1.1), alkaline phosphatase (EC 3.1.3.1), and  $\gamma$ -gluta-

myltransferase (EC 2.3.2.2).  $\alpha$ -Amylase was chosen because of the diversity of used substrates. Alkaline phosphatase and  $\gamma$ -glutamyltransferase were chosen for the differences in the nature and concentration of buffer and substrate. Furthermore, reference methods (1, 2) and certified reference materials (CRM) are available for alkaline phosphatase and  $\gamma$ -glutamyltransferase (3, 4) and are being developed for  $\alpha$ -amylase.

### Materials and Methods

#### General aspects

Five laboratories participated in this study. Each enzyme activity was determined independently in two laboratories, using automated procedures with the same reagent kits, at 30 and 37 °C, under routine conditions according to the procedure described by the manufacturer for each instrument. Each laboratory analysed specimens from its own patients (35 to 100 fresh sera, assayed in 6 to 10 series). Haemolysed, turbid or highly icteric samples were discarded. For each enzyme, samples with catalytic activity concentra-

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tions below the upper linearity limit of the methods were selected, taking care to ensure a homogeneous distribution of the values in the range tested. In each run, the corresponding CRM was also assayed (tab. 1). Two CRM (CRM 319 and CRM 371 for  $\gamma$ -glutamyltransferase and alkaline phosphatase, respectively) were purchased from the Community Bureau of Reference (BCR), Brussels. The  $\alpha$ -amylase candidate reference material (CRM 318) prepared under the BCR auspices was provided by Prof. F. J. Gella (Barcelona).

## Reagents

Various commercial kits of reagents were chosen because of their frequent use in Europe.

## $\alpha$ -Amylase catalytic activity concentration

$\alpha$ -Amylase was assayed with a kit containing maltotetraose (No. 63112, bioMérieux SA, Marcy-l'Étoile, France) and with kits containing different 4-nitrophenyloligosaccharides, i.e. 4-nitrophenylmalto- $\alpha$ -heptaoside (No. 568589, Boehringer-Mannheim GmbH, Mannheim, Germany), ethylidene-4-nitrophenylmalto- $\alpha$ -heptaoside (No. 1054671 & 1054643, Boehringer-Mannheim), benzylidene-4-nitrophenylmalto- $\beta$ -heptaoside (No. 63122, bioMérieux), 2-chloro-4-nitrophenylmalto- $\alpha$ -trioside (No. 63123, bioMérieux) and 2-chloro-4-nitrophenylmalto- $\beta$ -heptaoside (No. 12121, Merck, Darmstadt, Germany).

## Alkaline phosphatase catalytic activity concentration

Alkaline phosphatase was determined using 4-nitrophenylphosphate as substrate and two different buffer solutions. The first was diethanolamine in the kits from bioMérieux (No. 63609) and Boehringer-Mannheim (No. 415278) according to the DGKC recommended procedure. The second was 2-amino-2-methyl-1-propanol in the kits from bioMérieux (No. 63657), Biotrol (No. 03001, Biotrol, Paris, France), Boehringer-Mannheim (No. 396460) and Roche (No. 2813, Hoffmann-La Roche Ltd, Basle, Switzerland) according to the SFBC recommended procedure. Roche reagent in accordance with the IFCC recommendations (No. 0710571) was also selected in this study.

## $\gamma$ -Glutamyltransferase catalytic activity concentration

$\gamma$ -Glutamyltransferase was assayed with Boehringer-Mannheim and Merck kits containing  $\gamma$ -L-glutamyl-3-carboxy-4-nitroanilide (No. 158208 and No. 12191, respectively) and with bioMérieux

and Biotrol kits containing  $\gamma$ -L-glutamyl-4-nitroanilide as substrate (No. 63711 and No. A03027, respectively).

IFCC reference methods were also performed for alkaline phosphatase and  $\gamma$ -glutamyltransferase using home made reagents.

## Instruments

$\alpha$ -Amylase was run with a Cobas Mira analyser and a Cobas Fara analyser (Roche) in Laboratory 1 and Laboratory 2, respectively. Alkaline phosphatase was run with a Cobas Fara in Laboratory 2 and a Monarch apparatus (Instrumentation Laboratory, Milano, Italy) in Laboratory 3.  $\gamma$ -Glutamyltransferase was assayed with a Specific analyser (Kone Instrument Corporation, Espoo, Finland) in Laboratory 4 and a Cobas Mira in Laboratory 5.

For each enzyme and each method, instruments were set up according to the manufacturers' instructions. Thus, catalytic activity concentrations were first calculated using the factor based on the molecular absorption coefficient. Data were then recalculated using the corresponding CRM as calibrator.

## Data analysis

The mean of all results obtained for patient samples was calculated independently in each laboratory before and after calibration by the selected CRM. Certified values established (or currently determined) by the BCR were used for the calibration of each measurement procedure: 254 U/l for alkaline phosphatase using the IFCC method at 30 °C (1), 86.7 U/l for  $\gamma$ -glutamyltransferase using the IFCC method at 30 °C (2) and 549 U/l for  $\alpha$ -amylase using 2-chloro-4-nitrophenylmaltotrioside as substrate at 37 °C under the following conditions: 2-chloro-4-nitrophenylmaltotrioside, 2.25 mmol/l; potassium thiocyanate, 900 mmol/l; sodium chloride, 300 mmol/l; calcium chloride, 5 mmol/l; 2-morpholino-ethanesulphonic acid, 50 mmol/l, pH (37 °C) 6.28.

## Results

For each of the three enzymes assayed, results calculated with a theoretical calculation factor varied greatly according to the reagent used and temperature of measurement (figs. 1–3). Between-method dependency before calibration was more pronounced for  $\alpha$ -amylase and alkaline phosphatase than for  $\gamma$ -glutamyltransferase. Fig-

**Tab. 1** Some characteristics of the enzyme reference materials used for calibration.

Enzyme	Code	Origin	Form	Matrix	Certified value (method)	Reference
$\alpha$ -Amylase	CRM 318	Human pancreas	Pancreatic isoenzyme	Bovine albumin	549.0 $\pm$ 9.5 U/l (CI 4-NP G3, 37 °C)*	(5)
Alkaline phosphatase	CRM 371	Pig heart	Liver-bone-kidney isoform	Bovine albumin	254 $\pm$ 6 U/l (IFCC, 30 °C)	(6)
$\gamma$ -Glutamyltransferase	CRM 319	Pig kidney	Light subunit	Human albumin	117.8 $\pm$ 1.4 U/l (IFCC, 30 °C)	(7)

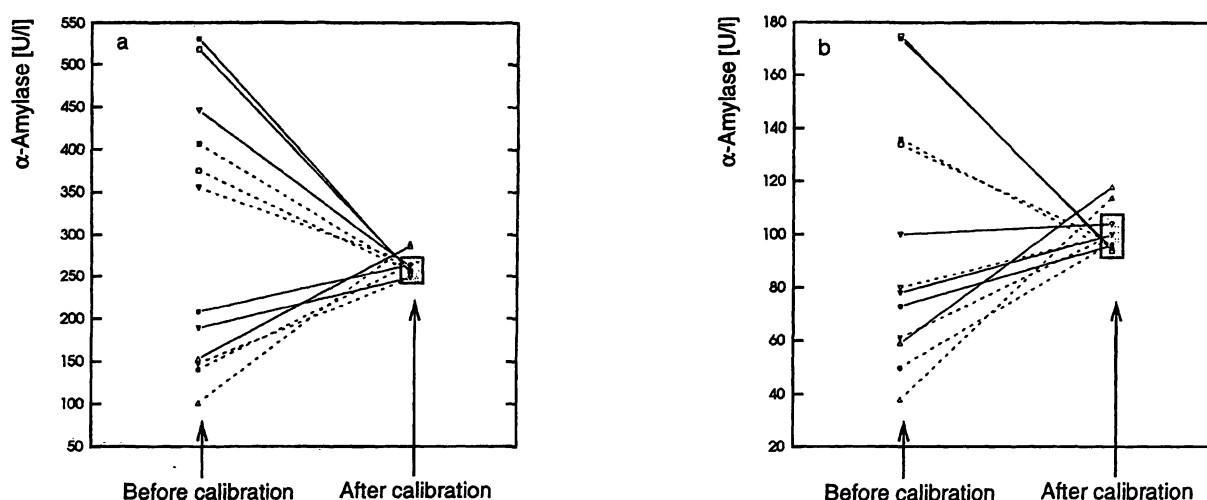
\* The final conditions for measurements were:

2-chloro-4-nitrophenyl- $\alpha$ -D-maltotriose (CI 4-NP G3)	2.25 mmol/l
potassium thiocyanate	900 mmol/l
sodium chloride	5 mmol/l
2-morpholinoethanesulphonic acid	50 mmol/l
pH	6.28 (37 °C)
volume fraction of the sample	0.0099

ures 1 to 3 indicate that the general patterns were similar in the two laboratories that assayed the same enzyme activity. Agreement between results in each participating laboratory was clearly improved by calibration. Nevertheless, the  $\alpha$ -amylase results obtained with maltotetraose after calibration still differ from those obtained with the other substrates employed in this study. This discrepancy was observed at both temperatures and in the two laboratories. For alkaline phosphatase, intermethod differences were not noticed. In contrast, after calibration, an intermethod disagreement was noted in the two laboratories for  $\gamma$ -glutamyltransferase results obtained with a kit containing a non-carboxylated substrate, compared with the results obtained with the other kits. Table

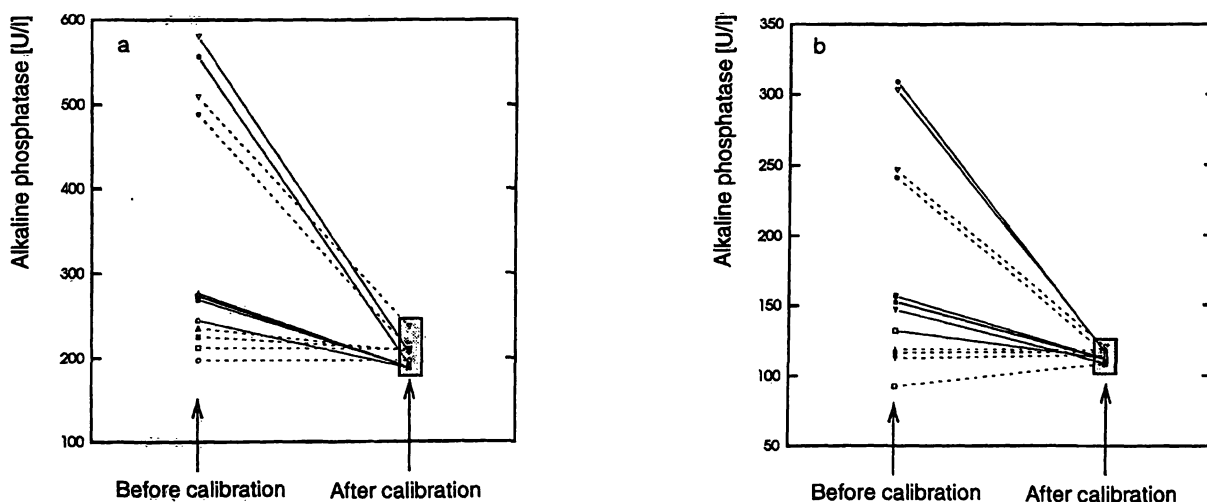
2 summarizes the calibration effect on the intermethod agreement. To quantify this effect, coefficients of variation were computerized for each series of mean values. A striking improvement was found for each enzyme and in each laboratory. After calibration, and after excluding maltotetraose results for  $\alpha$ -amylase and those obtained with a non-carboxylated substrate for  $\gamma$ -glutamyltransferase, all variation coefficients were less than 7.2%. Maximal relative variations linked to the measurement procedures were also considerably decreased by calibration (tab. 2).

The temperature effect, i. e. the ratio of activity concentration at 37 °C to that at 30 °C for the different reagents



**Fig. 1** Effect of calibration on the between-method coherency for  $\alpha$ -amylase activity. Each group corresponds to data obtained in one laboratory (a: lab. 1; b: lab. 2) at 30 °C (---) and 37 °C (—). Each sign corresponds to the mean enzyme catalytic concentration of patient samples for six measurement procedures using different

substrates:  $\Delta$  maltotetraose (bioMérieux);  $\square$  4-nitrophenylmaltoheptaoside (Boehringer);  $\square$  ethylidene-4-nitrophenylmaltoheptaoside (Boehringer);  $\bullet$  benzylidene-4-nitrophenylmaltoheptaoside (bioMérieux);  $\nabla$  2-chloro-4-nitrophenylmaltoheptaoside (Merck);  $\nabla$  2-chloro-4-nitrophenyl-maltotrioside (bioMérieux).



**Fig. 2** Effect of calibration on the between-method coherency for alkaline phosphatase activity. Each graph corresponds to data obtained in one laboratory (a: lab. 2; b: lab. 3) at 30 °C (---) and 37 °C (—). Each sign corresponds to the mean enzyme catalytic concentration of patient samples for eight measurement procedures, using as transphosphorylating buffer: 2-amino-2-methyl-1-propanol ac-

cording to IFCC recommendations ( $\circ$  home made reagent;  $\blacktriangle$  Roche), 2-amino-2-methyl-1-propanol according to SFBC recommendations ( $\square$  Boehringer;  $\nabla$  bioMérieux;  $\blacksquare$  Biotrol;  $\Delta$  Roche) and diethanolamine according to DGKC recommendations ( $\nabla$  Boehringer;  $\bullet$  bioMérieux).

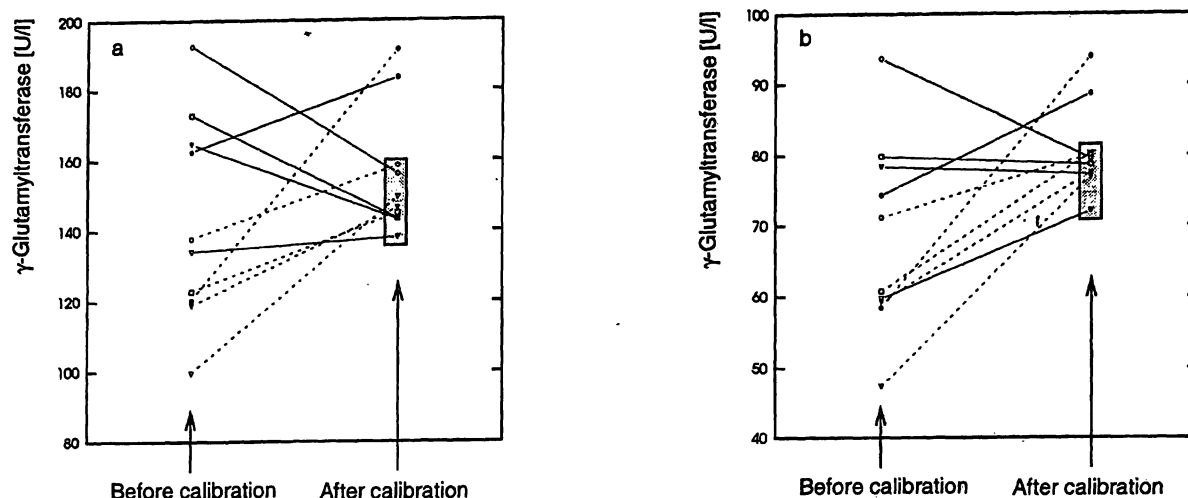


Fig. 3 Effect of calibration on the between-method coherency for  $\gamma$ -glutamyltransferase. Each graph corresponds to data obtained in one laboratory (a: lab. 4; b: lab. 5) at 30 °C (---) and 37 °C (—). Each sign corresponds to the mean enzyme catalytic concentration of patient samples for five measurement procedures using different

substrates at different concentrations.  $\gamma$ -L-glutamyl-3-carboxy-4-nitroanilide (○ IFCC method with a home made reagent; ▼ Boehringer; □ Merck) and  $\gamma$ -L-glutamyl-4-nitroanilide (● BioMérieux; ▽ Biotrol).

in two laboratories, is depicted in figure 4. Before calibration, it differed considerably with the used technique, especially for  $\alpha$ -amylase and alkaline phosphatase. The mean values and standard deviations calculated before and after calibration were respectively  $1.37 \pm 0.11$  and  $1.00 \pm 0.03$  for  $\alpha$ -amylase,  $1.28 \pm 0.08$  and  $0.94 \pm 0.03$  in the case of alkaline phosphatase and  $1.34 \pm 0.05$  and  $0.97 \pm 0.03$  for  $\gamma$ -glutamyltransferase. These results indicate that the differences between the temperature effects of techniques are reduced by calibration. Furthermore, the mean values of the ratios were close to 1.0 for the three enzymes after calibration.

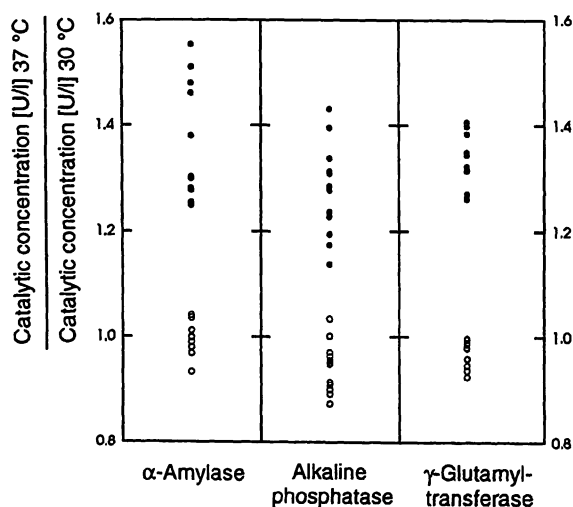


Fig. 4 Effect of reaction temperature on the activity concentration according to the used method of measurement. Each circle corresponds to a ratio for one procedure in one laboratory (● before calibration; ○ after calibration).

## Discussion

Three enzyme catalytic activity concentrations ( $\alpha$ -amylase, alkaline phosphatase,  $\gamma$ -glutamyltransferase) were chosen for testing the effect of calibration because marked intermethod discrepancies were observed in external quality control assessments. Measurement procedures were selected according to their current use in Europe. CRM from BCR were employed as calibrators in this study because previous studies have demonstrated that these materials can be purified and stabilized, without significant alteration of their catalytic properties (3–8). Furthermore, CRM were shown to exhibit the same catalytic properties as the corresponding enzyme in patient samples. Another report indicated that the calibration approach with CRM 319 allowed the transfer of accuracy from the  $\gamma$ -glutamyltransferase reference method to another method which differs e. g. in its reaction temperature (7).

The variety of routine methods in use in clinical laboratories causes problems in the interpretation and comparison of results.  $\alpha$ -Amylase and alkaline phosphatase results may vary by a factor of up to 3.5 according to the measurement procedure. This factor is close to 2 for  $\gamma$ -glutamyltransferase. Measurement temperature is one of the numerous factors with modify the results of enzyme activity measurements. However, our data obtained with a large number of patient samples covering a wide range of catalytic activity concentration indicate that the temperature effect on enzyme activities varies not only with the enzyme in question, but also with the measurement procedure (kit of reagents, apparatus and application on the analyser). This indicates that a con-

Tab. 2 Variability of results of enzyme activity measured by different procedures, before and after calibration by the corresponding CRM. Data were calculated from mean values of results obtained in each laboratory for its own patient samples, with each measurement procedure performed at 30 and 37 °C.

Enzyme (number of measurement procedures)	Laboratory	Results before calibration				Results after calibration			
		Mean [U/l]	Range [U/l]	Coefficient of variation [%]	Maximal relative variation <sup>d</sup>	Mean [U/l]	Range [U/l]	Coefficient of variation [%]	Maximal relative variation <sup>d</sup>
$\alpha$ -Amylase (5) <sup>a</sup>	Lab. 1	332	141–530	45.0	3.76	256	248–267	2.5	1.08
	Lab. 2	106	50–175	43.0	3.50	98	94–104	3.3	1.11
Alkaline phosphatase (8) <sup>b</sup>	Lab. 2	339	198–580	43.3	2.93	202	188–237	7.2	1.26
	Lab. 3	171	92–309	42.4	3.34	114	108–120	3.1	1.11
$\gamma$ -Glutamyltransferase (4) <sup>c</sup>	Lab. 4	143	100–193	22.0	1.93	147	138–158	4.6	1.14
	Lab. 5	69	47–94	22.0	2.00	78	72–80	3.4	1.11

<sup>a</sup> all measurement procedures, except the one using maltotetraose as substrate (bioMérieux)

<sup>b</sup> all measurement procedures

<sup>c</sup> all measurement procedures except the one using a non-carboxylated substrate (bioMérieux)

<sup>d</sup> maximal relative variation represents the ratio of range of mean values

stant temperature factor cannot be established for an enzyme. After calibration, the effect of reaction temperature was not apparent under our conditions. This is explained by the fact that temperature ratios were found to be very similar for patient samples and CRM for each measurement procedure (data not shown).

The results obtained in this study indicate that the calibration approach with an appropriate material (i.e. CRM) may be applied to the three tested enzymes with most of the reagents used under routine conditions. Thus, these methods constitute a set of measurement procedures in which numeric coherency of results between methods is strongly improved after calibration. For alkaline phosphatase, the seven kits of reagents and the IFCC procedure with home made reagent gave very similar results after calibration in both laboratories. Among the six kits chosen for  $\alpha$ -amylase, five appeared as a set of procedures in the two laboratories. After calibration, the kit using maltotetraose gave different results, as compared with the homogeneous group. In fact, it is well known that salivary and pancreatic isoforms of amylase do not transform maltotetraose at the same rate (9). The feasibility of calibration has already been reported by Gerhardt et al. with kits including blue starch polymer, amylose, blocked starch, oligosaccharides and 4-nitrophenyloligosaccharides, with the same observation concerning maltotetraose (10, 11). In other words, the maltotetraose procedure did not show the same analytical specificity as other procedures using a 4-nitrophenyloligosaccharide as substrate. In the case of  $\gamma$ -glutamyltransferase (five reagents), the discrepancy of a non-carboxylated substrate was also observed in both participant laboratories. It indicates that not only the reaction principle has to be taken into account when selecting a measurement procedure. Surprisingly,  $\gamma$ -glutamyltransferase which is the sole enzyme without isoenzyme heterogeneity, showed the least favourable effect of calibration. This observation is in agreement with a previous one concerning commutability studies (7). These authors found that various kits showed higher significant differences when comparing CRM 319 and human sera than those observed with home made reagents according to international and national (British, French and Scandinavian) recommendations. Some differences, such as concentrations of substrate or effectors and/or additives, may affect CRM and patient samples in a different manner. As this effect is not predictable, it has to be checked experimentally using a large number of patient samples and the candidate material(s) as enzyme calibrator. It has to be stressed that an enzyme preparation cannot be used as a calibrator without prior validation. Lipase activity provides a good example of how the use of calibrators is not sufficient to correct inter-assay disagreement. This enzyme is most often determined routinely by tech-

niques employing calibrators, but results may differ by a factor of up to 16 according to the measurement procedure (12). In fact, it has been shown that calibration with an appropriate material considerably improves inter-assay agreement (12). The importance of commutability of a material for intermethod calibration has also been recently discussed for amylase (11). The authors showed that materials with  $\alpha$ -amylase of non-human origin were not commutable with the enzyme in human sera and should not be used for calibration.

## Conclusion

It is possible to extensively reduce intermethod and interlaboratory discrepancies by the use of proper calibrators. Our data indicated that reaction temperature was

not a critical factor and that most of the tested reagents and procedures may be calibrated for the three enzymes tested. Nevertheless, at the present time, we have not yet found any commercial preparations that could be used as an enzyme multicalibrator. Secondary materials should be validated for each measurement procedure, using patient samples, prior to their routine use as enzyme calibrators.

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